

An Unusual Stereoelectronically-dictated, Quinone-mediated Dehydrogenation Implicated in Dienediaminoketone Nenitzescu Reactions

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For the first time dienediaminoketones (1-benzoyl-2-arylamino-4-dimethylaminobuta-1,3-dienes) have been used in a reaction with benzoquinone and as a result 2-dimethylamino-3-(1'-aryl-3'-benzoyl-5'-hydroxyindol-2'-yl)-5-hydroxybenzofurans have been synthesized.

The interaction of enaminoketones and *p*-benzoquinone with the formation of 5-hydroxy-benzofurans and -indoles (Nenitzescu reaction) has already been examined in detail.^{1,2} However, there are no data in the literature on the use of dienediaminoketones as the 'enamine' component in the Nenitzescu reaction. The synthesis and some properties of these compounds have been described by us recently,³ and their application in the Nenitzescu reaction is of undoubted interest since the course of their reactions with quinone is not evident – electrophilic attack can proceed on either the β - or the δ -position of the dienediamine system.

In the present work 1-benzoyl-2-arylamino-4-dimethylaminobuta-1,3-dienes **1a,b** were chosen as the starting compounds. These compounds **1a,b** react smoothly with benzoquinone **2** in acetic acid. It has been established experimentally that the reaction proceeds optimally when the ratio of reagents is 1:2 (in an equimolar ratio yields diminish and in the ratio 1:3 tar formation takes place).

2-Dimethylamino-3-(1'-aryl-3'-benzoyl-5'-hydroxyindol-2'-yl)-5-hydroxybenzofurans **3a,b**[†] were obtained as a result of the reaction between **1a,b** and **2**, Scheme 1: **3a**, yield 17%, m.p. 284–286 °C (AcOH);[‡] **3b**, yield 28%, m.p. 274–276 °C (AcOH).[§] In the EI mass spectra of **3a,b** intense molecular ion peaks are observed (see Table 1). The basic fragmentation of molecular ions proceeds according to a common scheme (Scheme 2) with elimination of NMe₂, CPh groups and

Table 1 Characteristic ion peaks in the spectra of compounds studied, *m/z* (relative intensity).^a

Ion	Compound			
	3a	3b	6	10
M ⁺	488(100)	518(100)	546(100)	532(100)
[M ⁺ – Me ₂ N] ⁺	444(58)	474(60)	502(30)	488(66)
[M – CPh] ⁺	383(25)	413(20)	441(15)	427(18)
A	326(60)	356(58)	370(77)	370(78)
B	196(35)	226(30)	240(24)	240(24)
PhCO ⁺	105(25)	105(25)	105(55)	105(31)
Ph ⁺	77(20)	77(20)	77(38)	77(20)
M ²⁺	244(16)	259(18)	273(30)	–

^aThe mass spectra of **3a,b** and **6** were recorded with a Finnigan SSQ-710 mass spectrometer using a direct inlet with ionizing potential 70 eV. The mass spectrum of **10** was obtained using a Finnigan MAT 112 instrument (50 eV).

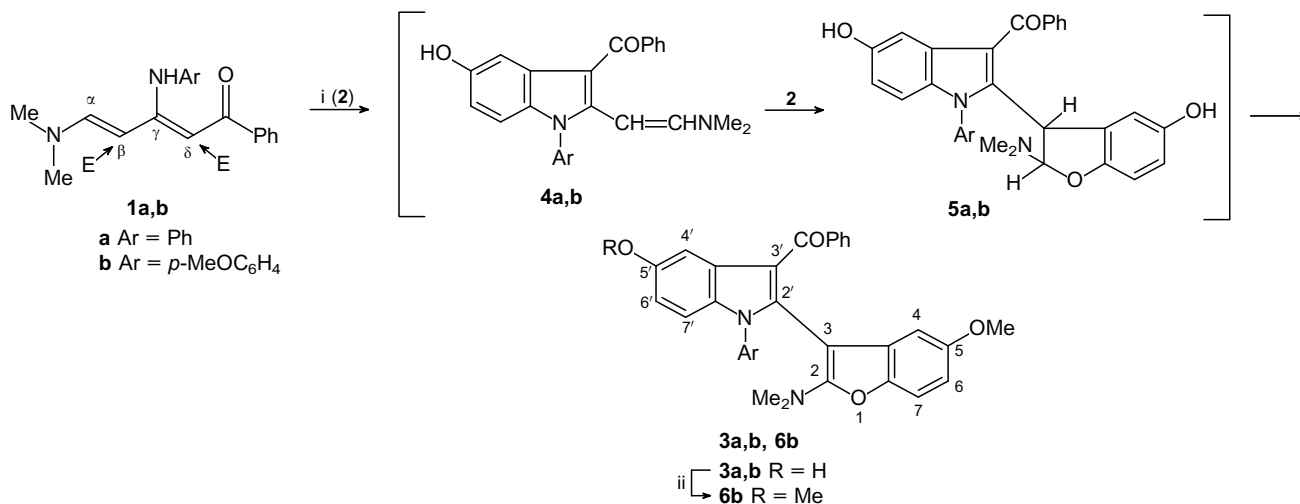
cleavage of the C(2')–C(3) bond between the indole and benzofuran rings. This cleavage is accompanied by migration of the CH₂ group from the NMe₂ group to the indole ring with quinolinium cation **A** formation. The latter fragmentation is connected with the elimination of a PhCOC≡C group to form ion **B**. It is also possible to observe in the spectra peaks due to discharged ions.

The Scheme 1 for **3a,b** formation shows that the Nenitzescu reaction proceeds unusually in this case and that cyclization to benzofuran derivatives is accompanied by dehydrogenation of C(2)–C(3) bonds with conservation of the Me₂N group at position 2. It is necessary to note that the reaction of **2** with enaminoketones usually proceeds with cleavage of the dimethylamine group resulting in formation of benzofuran without an amine-containing substituent at position 2.^{1,2} The above Scheme 1, in which quinone firstly

[†]All new compounds gave the expected IR, ¹H NMR and mass spectra and satisfactory elemental analysis data.

[‡]Spectroscopic data for **3a**: ¹H NMR ([²H₆]Me₂SO) δ 2.63 (s, 6H, NMe₂), 8.76 (s, 1H, OH), 9.16 (s, 1H, OH), 6.14–7.42 (m, 16H, arom. protons).

[§]Spectroscopic data for **3b**: ¹H NMR ([²H₆]Me₂SO) δ 2.67 (s, 6H, NMe₂), 3.75 (s, 3H, OMe), 6.20–7.48 (m, 15H, arom. protons), 8.81 (s, 1H, OH), 9.17 (s, 1H, OH).



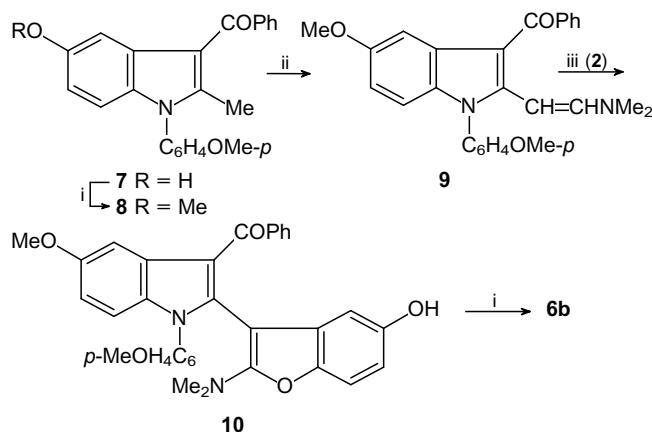
Scheme 1 Reagents and conditions: i, **1a,b** and **2**, AcOH, 15 h, 20 °C; ii, combination of Me₂SO₄:NaOH:triethylbenzylammonium chloride (TEBAC):**3a,b** = 6:3:0.2:2, CH₂Cl₂, H₂O, 4 h, 20 °C.

attacks the δ -position of the enamine, is based on results⁴ obtained from the reaction of **2** with nitrodienediamines in which the reaction proceeds at the δ -position. However, the interaction of nitrodienediamines with quinone leads to benzofuran (not to indole) derivatives. The reason for this difference is probably the stronger electron-withdrawing effect of the NO₂ group, which facilitates benzofuran cyclization (nucleophilic attack on the β -position relative to the nitro group) and hinders electrophilic attack of the second quinone molecule on the dimethylaminovinyl fragment of the 3-nitrobenzofuran derivative obtained.

The unusual (for the Nenitzescu reaction) conservation of the Me₂N group in the benzofuran ring is due to stereoelectronic factors: the protons at position 3 and the Me₂N group at position 2 in the intermediates **5a,b** are in an *S-syn*-position (according to Dreiding models). This hinders the cleavage of Me₂NH but facilitates the dehydrogenation stage (*S-anti* position of 2,3-protons), which probably proceeds by action of quinone **2**. The methylation of **3b** under phase-transfer catalysis⁵ conditions yields 74% of dimethoxy derivative **6b**, m.p. 180–181 °C (Pr'OH).[†] The final confirmation of the structures **3a,b** and **6b** is afforded by a supplementary synthesis: the methylation of 1-(*p*-methoxyphenyl)-2-methyl-5-hydroindole **7**⁶ gives 5-methoxy derivative **8**, yield 75%, m.p. 102–103 °C (heptane). Condensation of the latter with DMF diethylacetal yields 78% of indolyl enamine **9**, m.p. 147–149 °C (Pr'OH). Enamine **9** undergoes the Nenitzescu reaction with the formation of indolylbenzofuran **10**, yield 35%, m.p.

228–230 °C.* The methylation of **10** yields 87% of **6b**, which is identical with the sample from **3b** (see above).

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Scheme 3 Reagents and conditions: i, combination of Me₂SO₄:NaOH:TEBAC:**7,10** = 3:1.5:0.1:1, CH₂Cl₂, H₂O, 4 h, 20 °C; ii, Me₂NCH(OEt)₂, DMF, reflux for 25 h; iii, **9** and **2**, AcOH, 15 h, 20 °C.

References

- G. R. Allen, *Organic Reactions*, Wiley Interscience, New York, 1973, vol. 20, p. 337.
- V. G. Granik, V. M. Lyubchanskaya and T. I. Mukhanova, *Khim.-Farm. Zh.*, 1993, **6**, 37 [*Pharm. Chem. J. (Engl. Transl.)*, 1993, **6**, 413].
- V. M. Lyubchanskaya, T. I. Mukhanova, E. K. Panisheva, L. M. Alekseeva and V. G. Granik, *Mendeleev Commun.*, 1995, 24.
- V. M. Lyubchanskaya, L. M. Alekseeva and V. G. Granik, *Mendeleev Commun.*, 1995, 68.
- A. McKillop, J.-C. Fiaud and R. P. Hug, *Tetrahedron*, 1974, **30**, 1379.
- T. I. Mukhanova, L. M. Alekseeva, E. F. Kuleshova, Yu. N. Shejner and V. G. Granik, *Khim.-Farm. Zh.*, 1993, **2**, 60 [*Pharm. Chem. J. (Engl. Transl.)*, 1993, **2**, 136].

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[†] Spectroscopic data for **6b**: ¹H NMR ([²H₆]Me₂SO) δ 2.66 (s, 6H, NMe₂), 3.63, 3.72 and 3.79 (three s on 3H, OMe groups), 6.35 (q, 1H, 6-CH), 6.40 (d, 1H, 4-CH), 7.11 (d, 1H, 7'-CH), 7.58 (d, 1H, 4'-CH), 6.85–7.30 (m, 11H, N-C₆H₄, COPh, 7-CH, 6'-CH), mass spectra, see Table 1.

* Spectroscopic data for **10**: ¹H NMR ([²H₆]Me₂SO) δ 2.65 (s, 6H, NMe₂), 3.73 and 3.78 (two s on 3H, OMe groups), 6.93–7.31 (m, 9H, C₆H₄ and COPh), 6.23 (q, 1H, 6-CH), 6.26 (d, 1H, 4-CH), 6.77 (d, 1H, 7-CH), 6.89 (q, 1H, 6'-CH), 7.08 (d, 1H, 7'-CH), 7.58 (d, 1H, 4'-CH), mass spectra, see Table 1.